

Li-Hong Li, Xiang-Chao Zeng*
and Po-Run LiuDepartment of Chemistry, Jinan University,
Guangzhou, Guangdong 510632, People's
Republic of China

Correspondence e-mail: xczen@126.com

Key indicators

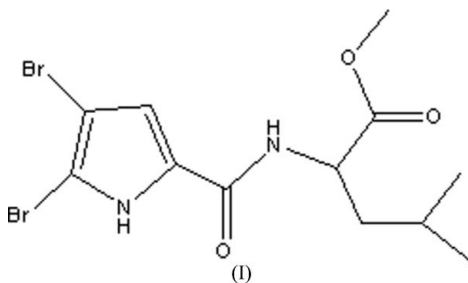
Single-crystal X-ray study
 $T = 293\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.020\text{ \AA}$
 R factor = 0.054
 wR factor = 0.173
Data-to-parameter ratio = 16.3For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.**(S)-Methyl 2-(4,5-dibromo-1*H*-pyrrole-
2-carboxamido)-4-methylpentanoate**

The title compound, $\text{C}_{12}\text{H}_{16}\text{Br}_2\text{N}_2\text{O}_3$, synthesized by the condensation of L-leucine methyl ester with 4,5-dibromo-2-trichloroacetylpyrrole at room temperature, crystallizes with two independent molecules in the asymmetric unit. Intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds link the molecules into ribbons extending along the b axis.

Received 18 April 2006
Accepted 25 April 2006

Comment

Pyrrole derivatives are well known as bioactive substances (Tasdemir *et al.*, 2002; Liu *et al.*, 2005). Examples are found in many marine organisms (Faulkner, 2001). In our search for bioactive compounds, a series of pyrrole(2-carbonyl)amino acid esters has been synthesized by the reaction of amino acid esters with 2-trichloroacetylpyrrole or brominated 2-trichloroacetylpyrroles. We report here the crystal structure of the title compound, (I).



Compound (I) crystallizes with two independent molecules in the asymmetric unit (Fig. 1). The bond lengths and angles in both molecules are unexceptional, being in good agreement with those observed in (*S*)-methyl 4-methyl-2-(1*H*-pyrrole-2-carboxamido)pentanoate (Zeng & Liu, 2005) and (*S*)-methyl 2-(4,5-dibromo-1*H*-pyrrole-2-carboxamido)-3-methylbutanoate (Zeng, 2006).

The crystal packing of (I) (Fig. 2) is stabilized by intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds (Table 1), which link the molecules into ribbons extending along the b axis.

Experimental

The hydrochloric acid salt of L-leucine methyl ester (0.91 g, 5 mmol) and 4,5-dibromo-2-trichloroacetylpyrrole (1.85 g, 5 mmol) were added to acetonitrile (12 ml), and then triethylamine (1.4 ml) was added dropwise. The mixture was stirred at room temperature for 9 h and then poured into water. After filtration, the precipitate was collected as a light-yellow solid. The impure product was dissolved in 95% ethanol at room temperature. Colourless orthorhombic crystals of (I) suitable for X-ray analysis (m.p. 437 K; 86.9% yield) grew over a period of one week when the solution was exposed to air.

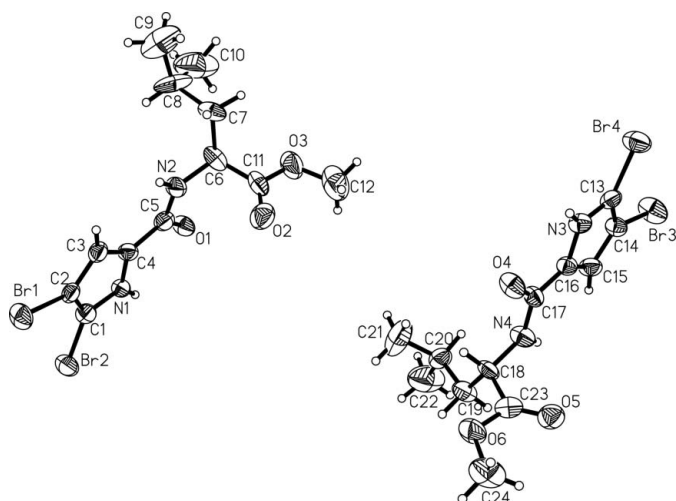


Figure 1
The two independent molecules of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

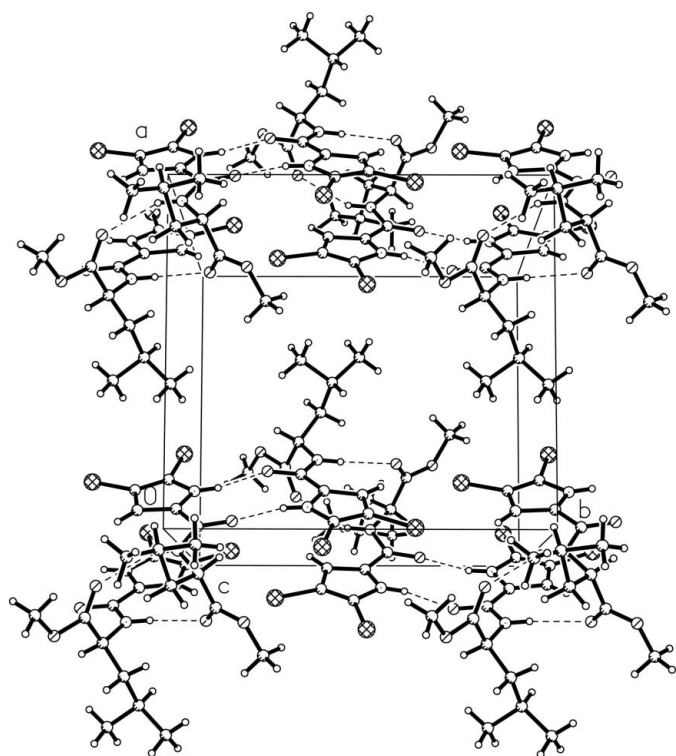


Figure 2
The crystal packing of (I), showing the intermolecular hydrogen bonds as dashed lines.

Crystal data

$C_{12}H_{16}Br_2N_2O_3$
 $M_r = 396.09$
 Orthorhombic, $P2_12_12_1$
 $a = 12.402$ (2) Å
 $b = 13.479$ (3) Å
 $c = 19.440$ (4) Å
 $V = 3249.8$ (11) Å³

$Z = 8$
 $D_x = 1.619$ Mg m⁻³
 Mo $K\alpha$ radiation
 $\mu = 4.99$ mm⁻¹
 $T = 293$ (2) K
 Block, colourless
 $0.44 \times 0.27 \times 0.19$ mm

Data collection

Bruker SMART 1K CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.217$, $T_{\max} = 0.451$
 (expected range = 0.187–0.387)

15481 measured reflections
 5694 independent reflections
 2490 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.086$
 $\theta_{\text{max}} = 25.0^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.055$
 $wR(F^2) = 0.173$
 $S = 1.02$
 5694 reflections
 349 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.052P)^2 + 3.8188P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.42$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.43$ e Å⁻³
 Absolute structure: Flack (1983),
 2473 Friedel pairs
 Flack parameter: 0.04 (2)

Table 1

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N4-H4 \cdots O2^i$	0.86	2.40	3.201 (16)	155
$N3-H3A \cdots O1^{ii}$	0.86	1.98	2.819 (14)	164
$N2-H2 \cdots O5^{iii}$	0.86	2.28	3.068 (17)	153
$N1-H1 \cdots O4^{iv}$	0.86	1.97	2.804 (13)	163

Symmetry codes: (i) $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $x - 1, y, z$; (iii) $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$; (iv) $x + 1, y, z$.

H atoms were positioned geometrically, with C–H = 0.93–0.98 Å and N–H = 0.86 Å, and refined using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2$ – 1.5 times U_{eq} of the parent atom.

Data collection: SMART (Bruker, 1999); cell refinement: SAINT-Plus (Bruker, 1999); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

The authors thank Dr Xiao-Long Feng and Dr Long Jiang, School of Chemistry and Chemical Engineering, Sun Yat-sen University, China, for their assistance in getting the crystal measured.

References

- Bruker (1997). SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruker (1999). SMART and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
 Faulkner, D. J. (2001). *Nat. Prod. Rep.* **18**, 1–49.
 Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
 Liu, J. F., Guo, S. P. & Jiang, B. (2005). *Chin. J. Org. Chem.* **25**, 788–799.
 Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
 Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
 Tasdemir, D., Mallon, R., Greenstein, M., Feldberg, L. R., Kim, S. C., Collins, K., Wojciechowicz, D., Mangalindan, G. C., Concepcion, G. P., Harper, M. K. & Ireland, C. M. (2002). *J. Med. Chem.* **45**, 529–532.
 Zeng, X.-C. (2006). *Acta Cryst.* **E62**, o288–o289.
 Zeng, X.-C. & Liu, P.-R. (2005). *Acta Cryst.* **E61**, o3726–o3727.